Synthesis of Pyrazines from Rhodium-Catalyzed Reaction of 2H-Azirines with N-Sulfonyl 1,2,3-Triazoles

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ABSTRACT: An efficient synthetic route to a wide range of trisubstituted pyrazines is developed from Rh-catalyzed reaction of 2H-azirines with N-sulfonyl-1,2,3-triazoles through the elimination of nitrogen molecule and arylsulfinic acid. The present reaction proceeds through formation of in situ generated dihydropyrazines.

■ INTRODUCTION

Development of a new synthetic method for azaheterocyclic compounds is highly significant in the investigation for new medicines, active pharmaceutical ingredients (API), and fine chemicals.¹ In particular, pyrazine is one of the most representative privileged azaheterocyclic scaffolds, which show cytostatic, [a](#page-6-0)ntifungal, and antitumor properties and are broadly present in flavorings and alarm pheromones (Figure 1).²

Accordingly, access to pyrazines from easily available starting materials is highly required. In general, the preparation of novel pyrazine scaffolds can be achieved by de novo synthesis from suitable starting materials. To date, some of the most usual de novo synthetic approaches described in the literature contain condensation of α -oximido carbonyl compounds with allylamines, 3 1,4-addition of 1,2-diamines to 1,2-diaza-1,3-butadienes,⁴ reductive condensation of α -nitro ketones with α amino [ke](#page-6-0)tones using electron transfer reagent,⁵ N−H insertion of Bo[c-a](#page-6-0)mino acid amides followed by acid-promoted cyclodehydration, 6 opening of epoxides with 1,2-a[mi](#page-6-0)no alcohols and Swern oxidation,⁷ Ru-catalyzed dehydrogenative coupling of β amino alcohols,⁸ and opening of 2H-azirines and dimerization.⁹ In addition, a [wi](#page-6-0)de range of pyrazines can be prepared by functionalizatio[n](#page-6-0) of a preformed pyrazine nucleus or cycl[o](#page-6-0)hexane derivatives having two nitrogen atoms at 1,4-position.¹⁰ However, some of these synthetic methods are restricted by their low yields, rigorous conditions, difficulties of prepari[ng](#page-6-0) unsymmetrical substituted pyrazines, or lack of substrate variation.

Over the last three years, the synthetic application of imino carbenes derived from N-sulfonyl-1,2,3-triazoles has been a very active area of research. 11 In this regard, we have intensively explored the utility of N-sulfonyl-1,2,3-triazoles as modular building blocks for the [pr](#page-6-0)eparation of a wide range of aza- and carbo-heterocyclic compounds¹² and C−H bond insertion of azulene. 13 In continuing studies, we envisaged that the Rhcatalyzed reaction of 2H-azirin[es](#page-7-0) with N-sulfonyl-1,2,3-triazoles would [allo](#page-7-0)w the formation of pyrazines. Herein we report an efficient method for the synthesis of a plethora of trisubstituted pyrazines from Rh-catalyzed reaction of 2H-azirines with Nsulfonyl-1,2,3-triazoles through the elimination of nitrogen molecule and arylsulfinic acid (Scheme 1).¹⁴

Scheme 1. Synthesis of Pyrazines throug[h](#page-7-0) Rhodium-Catalyzed Reaction

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■ RESULTS AND DISCUSSION

First, we commenced our studies with a variety of N-sulfonyl-4 phenyl-1,2,3-triazoles (2) generated from Cu-catalyzed $[3 + 2]$ cycloaddition of phenylacetylenes with sulfonyl azides.¹⁵ 4-Methoxyphenylsulfonyl azide was the best 1,3-dipolar reagent among several sulfonyl azides screened (methane-, isopro[pan](#page-7-0)e-, n-butane-, 4-trifluoromethylbenzene-, 4-methylbenzene-, and 4 methoxybenzene-sulfonyl azide). Next, Rh-catalyzed reaction of N-4-methoxybenzenesulfonyl-4-phenyl-1,2,3-triazole (2a) with ethyl 3-(4-nitrophenyl)-2H-azirine-2-carboxylate $(ia)^{16}$ was investigated (Table 1). Ethyl acetate was the best solvent

Table 1. Reaction Optimization^a

 a Reactions were carried out with 1a (0.2 mmol) and 2a (1.5 equiv) in solvent $(1.0 \text{ mL}, 0.2 \text{ M})$ at 120 °C for 16 h. b NMR yield using CH_2Br_2 as an internal standard. ^cIsolated yield of 3aa.

among several reaction media examined (dichloroethane, chloroform, toluene, chlorobenzene, n-hexane, cyclohexane, and ethyl acetate). A number of rhodium(II) catalysts were tested to reveal that $Rh_2(Oct)_4$ (2.0 mol %) was the catalyst of choice. The best result was obtained from a reaction of 1a (1.0 equiv, 0.2 mmol) with $2a$ (1.5 equiv) using $Rh_2(Oct)_4$ (2.0 mol %) in ethyl acetate at 120 °C for 16 h, producing ethyl 3-(4 nitrophenyl)-5-phenylpyrazine-2-carboxylate (3aa) in 70% isolated yield (entry 10). When the present reaction was conducted with $Rh_2(OAc)_4$ (2.0 mol %) at or below 100 °C, dihydropyrazines 4 and 5 were produced (entries 8 and 9). These results indicate that pyrazine 3aa is produced through elimination of arylsulfinic acid from dihydropyrazines 4 and 5. In addition, the present reaction did not proceed under thermal conditions without using Rh catalyst (entry 13). Optimization

of the stoichiometric ratio between starting materials 1a and 2a is described in the Supporting Information.

On the basis of the optimal reaction conditions, we next explored the subst[rate scope as well as th](#page-6-0)e functional group compatibility in the reaction with ethyl 3-(4-nitrophenyl)-2Hazirine-2-carboxylate (1a) (Scheme 2). Electronic variation of

Scheme 2. Rh-Catalyzed Synthesis of Pyrazines Using Various Triazoles^a

 a Reactions were carried out with 1a (0.2 mmol) and 2 (1.5 equiv) in solvent $(1.0 \text{ mL}, 0.2 \text{ M})$ at 120 °C for 16 h. b^2 (2.0 equiv) was used.

substituents at the aryl ring of N-sulfonyl triazoles (2) did not affect the reaction efficiency. For example, N-4-methoxybenzenesulfonyl-4-aryl-1,2,3-triazoles (2) having electron-donating 3 methyl, 4-methyl, 2-methoxy, and 3-methoxy groups on the phenyl ring underwent the Rh-catalyzed reactions, affording the desired pyrazines (3ab, 3ac, 3ad, and 3ae) in good yields ranging from 70% to 74%. The structure of 3ab was unambiguously determined by X-ray crystallography (see the Supporting Information).¹⁷ In addition, the reactions of substrates with electron-withdrawing 3-chloro, 4-chloro, 3 [bromo, and 4-bromo gro](#page-6-0)u[ps](#page-7-0) on the phenyl ring provided the cyclization products (3af, 3ag, 3ah, and 3ai) in moderate to good yields ranging from 51% to 81%. The tolerance of chloro

and bromo groups is especially meaningful, as further transformations of functional group are feasible. When a substrate having electron-withdrawing trifluoromethyl group was subjected to the Rh-catalyzed reaction, the product was formed in 68% yield. Substrate (2k) having 2-naphthyl group was also readily employed in the cyclization process. It was noteworthy that triazole bearing 3-thiophenyl group was also successfully applied to the current Rh-catalyzed reaction conditions, producing 3al in 54% yield. When cyclohexenylsubstituted triazole $(2m)$ was reacted with 1a in the presence of the rhodium catalyst, the desired pyrazine 3am was obtained in 51% yield. Triazole $(2n)$ having a *n*-butyl group at 4-position turned out to be compatible with the optimal reaction conditions, delivering the desired pyrazine 3an in 50% yield without β -hydride elimination.

With the success of the above reactions, we next investigated the effects of a number of substituents of ethyl 3-(4-aryl)-2Hazirine-2-carboxylate on cyclization (Scheme 3). Treatment of

Scheme 3. Rh-Catalyzed Synthesis of Pyrazines Using Various 2H-Azirines^a

 a Reactions were carried out with 1a (0.2 mmol) and 2 (1.5 equiv) in solvent $(1.0 \text{ mL}, 0.2 \text{ M})$ at 120 °C for 16 h. b^2 (2.0 equiv) was used.

triazole 2f with ethyl 3-(4-phenyl)-2H-azirine-2-carboxylate (1b) afforded ethyl 5-(3-chlorophenyl)-3-phenylpyrazine-2 carboxylate (3bf) in 52% yield. Electronic modification of substituents on the aryl ring of 2H-azirine 1 did not largely influence efficiency of the reaction. For instance, ethyl 3-(4 chlorophenyl)-2H-azirine-2-carboxylate (1c) was converted to the desired pyrazine 3cf in 73% yield. In addition, electronwithdrawing 3- and 4-bromo groups on the aryl ring afforded the cyclization products (3df and 3ef) in good yields. 2H-Azirine (1f) having 4-methoxycarbonyl group on the aryl ring worked equally well in the reaction with triazole 2f to produce the corresponding pyrazine 3ff in 70% yield. When the 2Hazirine (1g) bearing 2-naphthyl group was subjected to triazole 2f under the optimal conditions, Rh-catalyzed reaction smoothly took place to afford trisubstituted pyrazine 3gf in 58% yield. Unfortunately, 2H-azirines having electron-donating substituents such as methyl and methoxy groups on the aryl ring failed to prepare.

A proposed reaction pathway for the formation of pyrazine (3) from N-sulfonyl-1,2,3-triazole 2 and 2H-azirine 1 is shown in Scheme 4. First, a ring−chain tautomerization of triazole 2 followed by treatment of a rhodium catalyst provides α -imino rhodium [car](#page-3-0)benoid B along with evolution of nitrogen molecule. Addition of 2H-azirine 1 to the carbene center of B produces the rhodium-bound zwitterionic intermediate C. Then, a ring-opening reaction through the release of an electron pair from anionic rhodium of C provides dihydropyrazine 5 (pathway b). Finally, elimination of arylsulfinic acid from 5 produces pyrazine 3. 6π -Electrocyclization (pathway a) of 1,4-diazatriene D might be involved to the formation of dihydropyrazine 5. Because regioisomeric pyrazine H and pyrazine J are not observed from this reaction, an intramolecular hydrogen transfer (pathway d) and dimerization of nitrile ylide dipole (pathway c) can be ruled out.^{14,18} In fact, this postulation is supported by X-ray structure of pyrazine 3ab. In addition, tert-butyl 3-methyl-2H-azirine-2-car[boxyla](#page-7-0)te was not reacted with triazole, indicating that steric as well as electronic effects of substrate might be important for the formation of pyrazine.

■ CONCLUSION

In summary, we have developed an efficient synthetic method for a range of trisubstituted pyrazines from Rh-catalyzed reaction of 2H-azirines with N-sulfonyl-1,2,3-triazoles through the elimination of nitrogen molecule and arylsulfinic acid. The present reaction proceeds through formation of in situ generated dihydropyrazines.

EXPERIMENTAL SECTION

General. Reactions were carried out in oven-dried glassware under air atmosphere. $Rh_2(Oct)_4$, $Rh_2(OAc)_4$, $Rh_2(esp)_2$, $Rh_2(S-DOSP)_4$ were purchased and used as received. Commercial available reagents were used without purification. All reaction mixtures were stirred magnetically and were monitored by thin-layer chromatography using silica gel precoated glass plates, which were visualized with UV light and then, developed using either iodine or a solution of anisaldehyde. Flash column chromatography was carried out using silica gel (230− 400 mesh). $\rm ^1H$ NMR (400 MHz), $\rm ^{13}C(^{1}H)$ NMR (100 MHz), and $\rm ^{19}F$ NMR (377 MHz) spectra were recorded on NMR spectrometer. Deuterated chloroform was used as the solvent and chemical shift values (δ) are reported in parts per million relative to the residual signals of this solvent [δ 7.26 for ¹H (chloroform-d), δ 2.05 for ¹H (acetone- d_6), δ 77.2 for ¹³C{¹H} (chloroform- d), and δ 29.84, 206.26 for ${}^{13}\mathrm{C} \{ ^1\mathrm{H} \}$ (acetone- d_6)]. Infrared spectra were recorded on FT-IR spectrometer as either a thin film pressed between two sodium chloride plates or as a solid suspended in a potassium bromide disk. High resolution mass spectra (HRMS) were obtained by fast atom bombardment (FAB) using a double focusing magnetic sector mass spectrometer and electron impact (EI) ionization technique (magnetic sector−electric sector double focusing mass analyzer). Melting points were determined in open capillary tube.

Starting Materials of 2H-Azirines (1) and N-Sulfonyl-1,2,3- triazoles (2).12b,16c Ethyl 3-(4-Nitrophenyl)-2H-azirine-2-carboxylate (**1a**). Yellow solid, melting point: 84–89 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.4[4 \(d,](#page-7-0) J = 8.9 Hz, 2H), 8.10 (d, J = 8.8 Hz, 2H), 4.30–4.18 $(m, 2H)$, 2.97 (s, [1H](#page-7-0)), 1.29 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100) MHz, CDCl₃) δ 170.1, 158.6, 131.3, 130.4, 1280, 124.5, 61.7, 30.5, 14.2; IR (film) 2984, 1727, 1528, 1465, 1346, 1316, 1202, 752, 685 cm⁻¹; HRMS (FAB) [M + H]⁺ m/z calcd. for C₁₁H₁₁N₂O₄, 235.0719; found, 235.0717.

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Scheme 4. Proposed Mechanism

Ethyl 3-Phenyl-2H-azirine-2-carboxylate (**1b**). Yellow oil, ¹H NMR (400 MHz, CDCl₃) δ 7.90 (d, J = 8.1 Hz, 2H), 7.67-7.57 (m, 3H), 4.26−4.18 (m, 2H), 2.85 (s, 1H), 1.28 (t, J = 7.1 Hz, 3H); $^{13}C(^{1}H)$ NMR (100 MHz, CDCl₃) δ 171.7, 158.6, 133.9, 130.4, 129.3, 122.3, 61.3, 29.7, 14.2; IR (film) 2982, 1729, 1579, 1465, 1451, 1334, 1199, 762, 689 cm[−]¹ ; HRMS (FAB) [M + H]+ m/z calcd. for $C_{11}H_{12}NO_2$, 190.0868; found, 190.0868.

Ethyl 3-(4-Chlorophenyl)-2H-azirine-2-carboxylate (1c). Yellow solid, melting point: 57–62 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.84 $(d, J = 8.7 \text{ Hz}, 2H), 7.57 (d, J = 8.6 \text{ Hz}, 2H), 4.25–4.18 (m, 2H), 2.85$ (s, 1H), 1.28 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 171.4, 158.0, 140.4, 131.6, 129.8, 120.8, 61.4, 29.8, 14.2; IR (film) 2982, 1729, 1593, 1199, 835, 554 cm⁻¹; HRMS (FAB) [M + H]⁺ m/z calcd. for $C_{11}H_{11}CINO_2$, 224.0478; found, 224.0474.

Ethyl 3-(3-Bromophenyl)-2H-azirine-2-carboxylate (1d). Orange oil, ¹H NMR (400 MHz, CDCl₃) δ 8.03 (t, J = 1.6 Hz, 1H), 7.84 (dt, J $= 7.7, 1.3$ Hz, 1H), 7.77 (dq, J = 8.1, 1.0 Hz, 1H), 7.47 (t, J = 7.9 Hz, 1H), 4.26−4.17 (m, J = 7.1 Hz, 2H), 2.87 (s, 1H), 1.29 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 171.2, 158.1, 136.7, 132.9, 130.9, 128.8, 124.2, 123.2, 61.4, 30.0, 14.2; IR (film) 2981, 1729, 1566, 1472, 1199, 787, 679, 578 cm⁻¹; HRMS (FAB) [M + H]⁺ m/z calcd. for $C_{11}H_{11}BrNO_2$, 267.9973; found, 267.9975.

Ethyl 3-(4-Bromophenyl)-2H-azirine-2-carboxylate (1e). Orange solid, melting point, 65−70 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.75 (dd, J = 13.1, 8.8 Hz, 4H), 4.27−4.16 (m, 2H), 2.86 (s, 1H), 1.28 (t, J $= 7.1$ Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 171.4, 158.2, 132.8, 131.6, 129.0, 121.3, 61.4, 29.8, 14.2; IR (film) 2981, 1728, 1588, 1481, 1445, 1334, 1199, 831, 702 cm⁻¹; HRMS (FAB) [M + H]⁺ m/z calcd. for $C_{11}H_{11}BrNO_2$, 267.9973; found, 267.9972.

Ethyl 3-(4-(Methoxycarbonyl)phenyl)-2H-azirine-2-carboxylate (1**f**). Yellow oil, ¹H NMR (400 MHz, CDCl₃) δ 8.23 (d, J = 8.5 Hz, 2H), 7.97 (d, J = 8.6 Hz, 2H), 4.26−4.20 (m, 2H), 3.98 (s, 3H), 2.90 (s, 1H) 1.28 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 171.3, 165.8, 153.7, 134.8, 130.4, 130.3, 126.1, 61.5, 52.7, 30.1, 14.3; IR (film) 2984, 2954, 1726, 1571, 1335, 1199, 863, 769, 693 cm[−]¹ ; HRMS (FAB) $[M + H]^+ m/z$ calcd. for $C_{13}H_{14}NO_4$, 248.0923; found, 248.0920.

Ethyl 3-(Naphthalen-2-yl)-2H-azirine-2-carboxylate (1g). Yellow oil, ¹H NMR (400 MHz, CDCl₃) δ 8.32 (S, 1H), 8.03–7.92 (m, 4H), 7.69−7.59 (m, 2H), 4.29−4.18 (m, 2H), 2.94 (s, 1H), 1.29 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 171.8, 158.6, 135.8, 133.0, 132.7, 129.4, 129.17, 129.2, 128.1, 127.4, 124.8, 119.6, 61.4, 29.9, 14.3; IR (film) 2981, 1727, 1596, 1465, 1332, 1191, 863, 820, 750 cm⁻¹; HRMS (FAB) [M + H]⁺ m/z calcd. for C₁₅H₁₄NO₂, 240.1025; found, 240.1024.

1-((4-Methoxyphenyl)sulfonyl)-4-phenyl-1H-1,2,3-triazole (2a). White solid, melting point, 145−150 °C; ¹ H NMR (400 MHz, CDCl₃) δ 8.31 (s, 1H), 8.09 (d, J = 9.1 Hz, 2H), 7.84–7.82 (m, 2H), 7.46−7.41 (m, 2H), 7.39−7.35 (m, 1H), 7.04 (d, J = 9.1 Hz, 2H), 3.89 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.4, 147.3, 131.2, 129.1, 129.0, 128.9, 126.9, 126.1, 118.8, 115.1, 56.0; IR (film) 3090, 1590, 1268, 1166, 839, 768, 674 cm[−]¹ ; HRMS (EI) m/z calcd. for $C_{15}H_{13}N_3O_3S$, 315.0678; found, 315.0676.

1-((4-Methoxyphenyl)sulfonyl)-4-(p-tolyl)-1H-1,2,3-triazole (2b). White solid, melting point, 152−157 °C; ¹ H NMR (400 MHz, CDCl₃) δ 8.26 (s, 1H), 8.08 (d, J = 9.1 Hz, 2H), 7.71 (d, J = 8.2 Hz, 2H), 7.24 (d, J = 7.9 Hz, 2H), 7.04 (d, J = 9.1 Hz, 2H), 3.89 (s, 3H), 2.38 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.4, 147.4, 139.1, 131.2, 129.7, 127.1, 126.1, 126.0, 118.4, 115.1, 56.0, 21.4; IR (film)

3050, 1590, 1263, 1169, 1002, 980, 820, 770, 675 cm⁻¹; HRMS (EI) m/z calcd. for $C_{16}H_{15}N_3O_3S$, 329.0834; found, 329.0833.

1-((4-Methoxyphenyl)sulfonyl)-4-(m-tolyl)-1H-1,2,3-triazole (2c). White solid, melting point, 85−90 °C; ¹ H NMR (400 MHz, CDCl₃) δ 8.29 (s, 1H), 8.08 (d, J = 9.1 Hz, 2H), 7.67 (s, 1H), 7.61 $(d, J = 7.7 \text{ Hz}, 1\text{H}), 7.32 \text{ (t, } J = 7.7 \text{ Hz}, 1\text{H}), 7.19 \text{ (d, } J = 7.6 \text{ Hz}, 1\text{H}),$ 7.04 (d, J = 9.1 Hz, 2H), 3.89 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl3) δ 165.4, 147.4, 138.8, 131.2, 129.8, 128.9, 128.8, 127.0, 126.7, 123.2, 118.8, 115.1, 56.0, 21.4; IR (film) 2916, 1592, 1268, 1197, 1166, 1091, 1006, 836, 785, 680 cm[−]¹ ; HRMS (EI) m/z calcd. for $C_{16}H_{15}N_3O_3S$, 329.0834; found, 329.0835.

4-(2-Methoxyphenyl)-1-((4-methoxyphenyl)sulfonyl)-1H-1,2,3-triazole (2d). White solid, melting point, 110-115 °;, ¹H NMR (400 MHz, CDCl₃) δ 8.57 (s, 1H), 8.32 (dd, J = 7.8, J = 1.7 Hz, 1H), 8.09 $(d, J = 9.0 \text{ Hz}, 2H), 7.34 \text{ (td, } J = 7.8, J = 1.7 \text{ Hz}, 1H), 7.09 - 7.00 \text{ (m, }$ 4H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.2, 156.0, 142.7, 131.1, 130.0, 128.0, 127.4, 122.0, 121.0, 117.8, 115.0, 110.8, 56.0, 55.5; IR (film): 3076, 1592, 1267, 1022, 835, 755, 678 cm[−]¹ ; HRMS (EI) m/z calcd. for $C_{16}H_{15}N_3O_4S$, 345.0783; found, 345.0785.

4-(3-Methoxyphenyl)-1-((4-methoxyphenyl)sulfonyl)-1H-1,2,3-triazole (2e). White solid, melting point, 76–81 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.30 (s, 1H), 8.09 (d, J = 9.0 Hz, 2H), 7.42–7.31 (m, 3H), 7.04 (d, $J = 9.0$ Hz, 2H), 6.92 (dt, $J = 7.4$, 2.2 Hz, 1H), 3.89 (s, 3H), 3.86 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.4, 160.1, 147.2, 131.2, 130.2, 130.1, 126.9, 119.0, 118.4, 115.1, 111.1, 77.3, 56.0, 55.4; IR (film) 2943, 1592, 1269, 1044, 1024, 835, 783, 678 cm⁻¹; HRMS (EI) m/z calcd. for $C_{16}H_{15}N_3O_4S$, 345.0783; found, 345.0786.

4-(3-Chlorophenyl)-1-((4-methoxyphenyl)sulfonyl)-1H-1,2,3-triazole (2f). White solid, melting point, 132–137 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.32 (s, 1H), 8.09 (d, J = 9.1 Hz, 2H), 7.84–7.83 (m, 1H), 7.71 (dt, J = 7.0, 1.7, Hz, 1H), 7.39−7.33 (m, 2H), 7.05 (d, J = 9.1 Hz, 2H), 3.89 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.5, 146.0, 135.0, 131.3, 130.8, 130.3, 129.1, 126.8, 126.1, 124.1, 119.3, 115.2, 56.0; IR (film) 3143, 1292, 1269, 1022, 963, 767, 676, 586 cm⁻¹; HRMS (EI) m/z calcd. for C₁₅H₁₂ClN₃O₃S, 349.0288; found, 349.0286.

4-(4-Chlorophenyl)-1-((4-methoxyphenyl)sulfonyl)-1H-1,2,3-triazole (2g). White solid, melting point, 129–134 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.30 (s, 1H), 8.09 (d, J = 9.1 Hz, 2H), 7.77 (d, J = 8.6 Hz, 2H), 7.41 (d, J = 8.6 Hz, 2H), 7.05 (d, J = 9.1 Hz, 2H), 3.89 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.5, 146.2, 134.9, 131.3, 129.3, 127.5, 127.3, 126.8, 118.9, 115.2, 56.0; IR (film) 3095, 1591, 1267, 1022, 836, 781, 679, 591 cm⁻¹; HRMS (EI) m/z calcd. for $C_{15}H_{12}CN_3O_3S$, 349.0288; found, 349.0284.

4-(3-Bromophenyl)-1-((4-methoxyphenyl)sulfonyl)-1H-1,2,3-triazole (2h). White solid, melting point, 103–108 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.32 (s, 1H), 8.09 (d, J = 9.1 Hz, 2H), 7.99 (t, J = 1.7 Hz, 1H), 7.78−7.75 (m, 1H), 7.50 (ddd, J = 8.0, 1.9, 1.0 Hz, 1H), 7.31 $(t, J = 7.9$ Hz, 1H), 7.05 (d, $J = 9.1$ Hz, 2H), 3.90 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.6, 145.9, 132.1, 131.4, 131.1, 130.7, 129.1, 126.8, 124.7, 123.2, 119.4, 115.3, 56.1; IR (film) 2944, 1592, 1269, 1024, 835, 784, 676, 589 cm⁻¹; HRMS (EI) m/z calcd. for $C_{15}H_{12}BrN_3O_3S$, 392.9783; found, 392.9783.

4-(4-Bromophenyl)-1-((4-methoxyphenyl)sulfonyl)-1H-1,2,3-triazole (2i). White solid, melting point, 155–160 °C; ¹H NMR (400 MHz, acetone- d_6) δ 9.07 (s, 1H), 8.15 (d, J = 9.0 Hz, 2H), 7.94 (d, J = 8.4 Hz, 2H), 7.68 (d, J = 8.4 Hz, 2H), 7.26 (d, J = 9.0 Hz, 2H), 3.96 (s, 3H); ¹³C{¹H} NMR (100 MHz, acetone- d_6) δ 166.6, 146.9, 132.9, 132.0, 129.5, 128.6, 127.7, 123.2, 121.4, 116.3, 56.6; IR (film) 2976, 1591, 1267, 1197, 1019, 837, 786, 679, 593 cm⁻¹; HRMS (EI) m/z calcd. for $C_{15}H_{12}BrN_3O_3S$, 392.9783; found, 392.9784.

1-((4-Methoxyphenyl)sulfonyl)-4-(4-(trifluoromethyl)phenyl)-1H-1,2,3-triazole (2j). White solid, melting point, 133–138 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 8.39 (s, 1H), 8.10 (d, J = 9.1 Hz, 2H), 7.96 (d, J $= 8.0$ Hz, 2H), 7.70 (d, J = 8.2 Hz, 2H), 7.06 (d, J = 9.1 Hz, 2H), 3.90 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.6, 145.8, 131.4, 130.9 (q, J = 32.5 Hz), 126.2, 126.02, 126.01 (q, J = 4.2 Hz), 125.98, 123.9 (q, J = 272.2 Hz), 119.7, 115.2, 56.0; 19F NMR (377 MHz, CDCl3) δ −62.73; IR (film) 3106, 1596, 1394, 1326, 1270, 1164,

1024, 838, 829, 679 cm⁻¹; HRMS (EI) *m/z* calcd. for C₁₆H₁₂F₃N₃O₃S, 383.0551; found, 383.0549.

1-((4-Methoxyphenyl)sulfonyl)-4-(naphthalen-2-yl)-1H-1,2,3-triazole (2k). White solid, melting point, 130–135 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.42 (s, 1H), 8.37 (s, 1H), 8.11 (d, J = 9.1 Hz, 2H), 7.92−7.84 (m, 4H), 7.54−7.49 (m, 2H), 7.05 (d, J = 9.1 Hz, 2H), 3.90 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.4, 147.3, 133.5, 133.4, 131.3, 128.8, 128.3, 127.8, 127.0, 126.7, 126.67, 126.2, 125.3, 123.6, 125.3, 123.6, 119.1, 115.1, 56.0; IR (film) 3144, 1592, 1268, 1019, 804, 751, 676 cm⁻¹; HRMS (EI) m/z calcd. for C₁₉H₁₅N₃O₃S, 365.0834; found, 365.0832.

1-((4-Methoxyphenyl)sulfonyl)-4-(thiophen-3-yl)-1H-1,2,3-triazole (2l). White solid, melting point, 150–155 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.20 (s, 1H), 8.08–8.05 (m, 2H), 7.76–7.75 (m, 1H), 7.43−7.39 (m, 2H), 7.05−7.02 (m, 2H), 3.89 (s, 3H); 13C{1 H} NMR $(100 \text{ MHz}, \text{CDCl}_3)$ δ 165.4, 143.4, 131.2, 130.1, 126.9, 126.8, 125.7, 122.7, 118.5, 115.1, 56.0; IR (film) 3098, 1590, 1268, 1006, 837, 785, 676 cm⁻¹; HRMS (EI) *m/z* calcd. for $C_{13}H_{11}N_3O_3S_2$, 321.0242; found, 321.0242.

4-(Cyclohex-1-en-1-yl)-1-((4-methoxyphenyl)sulfonyl)-1H-1,2,3 t*riazole (2m)*. White solid, melting point, 96–100 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.03 (d, J = 9.1 Hz 2H), 7.89 (s, 1H), 7.01 (d, J = 9.1 Hz, 2H), 6.67−6.64 (m, 1H), 3.88 (s, 3H), 2.33−2.29 (m, 2H), 2.22− 2.17 (m, 2H), 1.78−1.72 (m, 2H), 1.68−1.62 (m, 2H); 13C{1 H} NMR $(100 \text{ MHz}, \text{CDCl}_3)$ δ 165.2, 148.9, 131.0, 127.5, 127.2, 125.8, 117.3, 115.0, 56.0, 26.2, 25.3, 22.2, 22.0; IR (film) 2931, 2858, 1592, 1268, 1020, 966, 835, 804, 676 cm[−]¹ ; HRMS (EI) m/z calcd. for $C_{15}H_{17}N_3O_3S$, 319.0991; found, 319.0992.

4-Butyl-1-((4-methoxyphenyl)sulfonyl)-1H-1,2,3-triazole (2n). White solid, melting point, 75−80 °C; ¹ H NMR (400 MHz, CDCl₃) δ 8.04 (d, J = 9.1 Hz, 2H), 7.83 (s, 1H), 7.02 (d, J = 9.1 Hz, 2H), 3.89 (s, 3H), 2.71 (t, J = 15.4 Hz, 2H), 1.68–1.60 (m, 2H), 1.39−1.34 (m, 2H), 0.92 (t, J = 7.4 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.2, 148.2, 131.0, 127.2, 120.2, 115.0, 55.9, 31.0, 25.1, 22.2, 13.7; IR (film) 2957, 2932, 1593, 1462, 1269, 1013, 835, 715, 676 cm⁻¹; HRMS (EI) m/z calcd. for C₁₃H₁₇N₃O₃S, 295.0991; found, 295.0993.

General Procedure for the Preparation of Pyrazines (3). To a screw-top V-vial were added 2H-azirine derivatives (2a, 0.2 mmol), triazole derivatives (1a, 0.3 mmol), and $Rh_2(Oct)_4$ (3.1 mg, 0.004 mmol) in EtOAc (1.0 mL). The resulting mixture was stirred at 120 °C for 16 h. After Celite filtration and evaporation of the solvents in vacuo, the crude product was purified by column chromatography on silica gel (EtOAc:Hx = 1:5) to yield 3aa (48.9 mg, 70%) as a white solid.

Ethyl 3-(4-Nitrophenyl)-5-phenylpyrazine-2-carboxylate (3aa). Yield 48.9 mg (70%), $R_f = 0.26$ (EtOAc:Hx = 1:5); White solid, melting point, 137–142 ^{o'}C; ¹H NMR (400 MHz, CDCl₃) δ 9.12 (s, 1H), 8.37 (d, J = 8.9 Hz, 2H), 8.16−8.14 (m, 2H), 7.87 (d, J = 8.9 Hz, 2H), 7.58−7.55 (m, 3H), 4.36 (q, J = 7.1 Hz, 2H), 1.27 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.6, 153.0, 151.3, 148.4, 144.0, 141.8, 139.8, 134.9, 131.1, 129.9, 129.3, 127.5, 123.5, 62.5, 13.9; IR (film) 3059, 2978, 1730, 1520, 1437, 1347, 1138, 1012, 853 cm[−]¹ ; HRMS (EI) m/z calcd for $C_{19}H_{15}N_3O_4$, 349.1063; found, 349.1060.

Ethyl 3-(4-Nitrophenyl)-5-(p-tolyl)pyrazine-2-carboxylate (3ab). Yield 51.6 mg (71%), $R_f = 0.2$ (EtOAc:Hx = 1:5); white solid, melting point, 131−136 °C; ¹H NMR (400 MHz, CDCl₃) δ 9.09 (s, 1H), 8.36 $(d, J = 8.8 \text{ Hz}, 2H), 8.05 (d, J = 8.2 \text{ Hz}, 2H), 7.86 (d, J = 8.9 \text{ Hz}, 2H),$ 7.36 (d, J = 8.0 Hz, 2H), 4.35 (q, J = 7.1 Hz, 2H), 2.45 (s, 3H), 1.27 (t, $J = 7.1$ Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.6, 153.1, 151.4, 148.3, 144.2, 141.7, 141.3, 139.5, 132.1, 130.1, 129.9, 127.4, 123.5, 62.5, 21.5, 13.9; IR (film) 3056, 2982, 1728, 1521, 1443, 1348, 1139, 1014, 853 cm⁻¹; HRMS (EI) m/z calcd for C₂₀H₁₇N₃O₄, 363.1219; found, 363.1216.

Ethyl 3-(4-Nitrophenyl)-5-(m-tolyl)pyrazine-2-carboxylate (3ac). Yield 53.8 mg (74%), $R_f = 0.24$ (EtOAc:Hx = 1:5); white solid, melting point, 105−110 °C; ¹H NMR (400 MHz, CDCl₃) δ 9.10 (s, 1H), 8.37 (d, J = 8.9 Hz, 2H), 7.95−7.92 (m, 2H), 7.87 (d, J = 8.9 Hz, 2H), 7.45 (t, J = 7.6 Hz, 1H), 7.37 (d, J = 7.6 Hz, 1H), 4.36 (q, J = 7.1 Hz, 2H), 2.48 (s, 3H), 1.27 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.6, 153.2, 151.4, 148.4, 144.1, 141.6, 139.9, 139.2, 134.8, 132.0, 129.9, 129.2, 128.1, 124.7, 123.5, 62.5, 21.6, 13.9; IR (film) 3053, 2982, 1731, 1521, 1443, 1348, 1137, 1014, 853 cm[−]¹ ; HRMS (EI) m/z calcd for $C_{20}H_{17}N_3O_4$, 363.1219; found, 363.1217.

Ethyl 5-(2-Methoxyphenyl)-3-(4-nitrophenyl)pyrazine-2-carboxy*late* (3*ad*). Yield 53.2 mg (70%), $R_f = 0.14$ (EtOAc:Hx = 1:5); pale yellow solid, melting point, 135−140 °C; ¹H NMR (400 MHz, CDCl₃) δ 9.33 (s, 1H), 8.35 (d, J = 8.8 Hz, 2H), 7.99 (dd, J = 7.7, 1.8 Hz, 1H), 7.85 (d, J = 8.9 Hz, 2H), 7.50 (ddd, J = 8.3, 7.4, 1.7 Hz, 1H), 7.13 (td, $J = 7.5$, 0.9 Hz, 1H), 7.07 (d, $J = 8.3$ Hz, 1H), 4.36 (q, $J = 7.1$ Hz, 2H), 3.95 (s, 3H), 1.28 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.7, 157.7, 152.4, 151.3, 148.2, 144.4, 144.3, 140.7, 132.3, 131.5, 129.9, 124.3, 123.5, 121.5, 111.6, 62.4, 55.7, 14.0; IR (film) 3078, 2980, 1731, 1520, 1436, 1347, 1139, 1019, 853 cm[−]¹ ; HRMS (EI) m/z calcd for $C_{20}H_{17}N_3O_5$, 379.1168; found, 379.1169.

Ethyl 5-(3-Methoxyphenyl)-3-(4-nitrophenyl)pyrazine-2-carboxy*late (3ae).* Yield 53.0 mg (70%), $R_f = 0.2$ (EtOAc:Hx = 1:5); pale yellow solid, melting point, 155−160 °C; ¹ H NMR (400 MHz, CDCl₃) δ 9.10 (s, 1H), 8.36 (d, J = 8.8 Hz, 2H), 7.87 (d, J = 8.8 Hz, 2H), 7.71−7.69 (m, 2H), 7.47 (t, J = 8.2 Hz, 1H), 7.10−7.08 (m, 1H), 4.36 (q, J = 7.1 Hz, 2H), 3.90 (s, 3H), 1.27 (t, J = 7.1 Hz, 3H);
¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.5, 160.4, 152.8, 151.3, 148.4, 144.0, 141.9, 139.9, 136.2, 130.4, 129.9, 123.5, 119.8, 116.6, 113.1, 62.5, 55.5, 13.9; IR (film) 3075, 2938, 1728, 1520, 1459, 1348, 1136, 1028, 853 cm⁻¹; HRMS (EI) *m/z* calcd for C₂₀H₁₇N₃O₅, 379.1168; found, 379.1165.

Ethyl 5-(3-Chlorophenyl)-3-(4-nitrophenyl)pyrazine-2-carboxy*late (3af)*. Yield 57.5 mg (75%), $R_f = 0.23$ (EtOAc:Hx = 1:5); pale yellow solid, melting point, 130−135 °C; ¹ H NMR (400 MHz, CDCl₃) δ 9.10 (s, 1H), 8.37 (d, J = 8.9 Hz, 2H), 8.16–8.15 (m, 1H), 8.03−8.00 (m, 1H), 7.87 (d, J = 8.8 Hz, 2H), 7.54−7.48 (m, 2H), 4.37 $(q, J = 7.1 \text{ Hz}, 2H)$, 1.27 $(t, J = 7.1 \text{ Hz}, 3H)$; ¹³C{¹H} NMR (100) MHz, CDCl₃) δ 165.4, 151.6, 151.4, 148.5, 143.6, 142.5, 139.7, 136.6, 135.5, 131.1, 130.6, 129.9, 127.6, 125.5, 123.6, 62.6, 13.9; IR (film) 3053, 2986, 1731, 1525, 1421, 1349, 1143, 1014, 854 cm⁻¹; HRMS (EI) m/z calcd for $C_{19}H_{14}C/N_3O_4$, 383.0673; found, 383.0674.

Ethyl 5-(4-Chlorophenyl)-3-(4-nitrophenyl)pyrazine-2-carboxy*late* (3*ag*). Yield 39.1 mg (51%), $R_f = 0.23$ (EtOAc:Hx = 1:5); pale yellow solid, melting point, 178−183 °C; ¹ H NMR (400 MHz, CDCl₃) δ 9.09 (s, 1H), 8.37 (d, J = 8.8 Hz, 2H), 8.10 (d, J = 8.7 Hz, 2H), 7.86 (d, $J = 8.8$ Hz, 2H), 7.54 (d, $J = 8.7$ Hz, 2H), 4.36 (q, $J = 7.1$ Hz, 2H), 1.27 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.4, 151.8, 151.4, 148.5, 143.8, 142.1, 139.5, 137.6, 133.3, 129.9, 129.6, 128.7, 123.6, 62.6, 13.9; IR (film) 3058, 2984, 1721, 1519, 1442, 1349, 1140, 1010, 853, 516 cm⁻¹; HRMS (EI) m/z calcd for $C_{19}H_{14}ClN_3O_4$, 383.0673; found, 383.0675.

Ethyl 5-(3-Bromophenyl)-3-(4-nitrophenyl)pyrazine-2-carboxylate (3ah). Yield 69.2 mg (81%), $R_f = 0.2$ (EtOAc:Hx = 1:5); pale yellow solid, melting point, 150−155 °C; ¹ H NMR (400 MHz, CDCl₃) δ 9.09 (s, 1H), 8.38 (d, J = 8.8 Hz, 2H), 8.30 (t, J = 1.8 Hz, 1H), 8.06 (ddd, J = 7.9, 1.5, 1.0 Hz, 1H), 7.87 (d, J = 8.8 Hz, 2H), 7.68 $(\text{ddd}, J = 8.0, 1.9, 0.9 \text{ Hz}, 1H), 7.44 \text{ (t, } J = 7.9 \text{ Hz}, 1H), 4.37 \text{ (q, } J = 7.1$ Hz, 2H), 1.27 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.4, 151.5, 151.4, 148.5, 143.6, 142.5, 139.7, 136.8, 134.1, 130.8, 130.5, 129.9, 125.9, 123.6, 62.6, 13.9; IR (film) 3068, 2983, 1731, 1520, 1443, 1348, 1141, 1065, 853 cm⁻¹; HRMS (EI) m/z calcd for $C_{19}H_{14}BrN_3O_4$, 427.0168; found, 427.0166.

Ethyl 5-(4-Bromophenyl)-3-(4-nitrophenyl)pyrazine-2-carboxylate (3ai). Yield 45.3 mg (53%), $R_f = 0.2$ (EtOAc:Hx = 1:5); pale yellow solid, melting point, 192−197 °C; ¹ H NMR (400 MHz, CDCl₃) δ 9.09 (s, 1H), 8.37 (d, J = 8.9 Hz, 2H), 8.03 (d, J = 8.7 Hz, 2H), 7.86 (d, J = 8.9 Hz, 2H), 7.69 (d, J = 8.7 Hz, 2H), 4.36 (q, J = 7.1 Hz, 2H), 1.27 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.4, 151.9, 151.4, 148.5, 143.7, 142.1, 139.4, 133.7, 132.6, 129.9, 128.9, 126.1, 123.6, 62.6, 13.9; IR (film) 3058, 2983, 1720, 1520, 1441, 1348, 1139, 1006, 853 cm[−]¹ ; HRMS (EI) m/z calcd for $C_{19}H_{14}BrN_3O_4$, 427.0168; found, 427.0164.

Ethyl 3-(4-Nitrophenyl)-5-(4-(trifluoromethyl)phenyl)pyrazine-2 *carboxylate (3aj).* Yield 56.5 mg (68%), $R_f = 0.23$ (EtOAc:Hx = 1:5); pale yellow solid, melting point, 165−170 °C; ¹ H NMR (400 MHz, CDCl₃) δ 9.16 (s, 1H), 8.38 (d, J = 8.9 Hz, 2H), 8.27 (d, J = 8.1) Hz, 2H), 7.88 (d, J = 8.8 Hz, 2H), 7.82 (d, J = 8.2 Hz, 2H), 4.38 (q, J = 7.1 Hz, 2H), 1.28 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl3) δ 165.4, 151.5, 151.4, 148.5, 143.5, 142.8, 139.9, 138.2, 132.8 $(q, J = 32.8 \text{ Hz})$, 129.9, 127.9, 126.2 $(q, J = 3.7 \text{ Hz})$, 123.8 $(q, J =$ 272.9 Hz), 123.6, 62.7, 13.9; ¹⁹F NMR (377 MHz, CDCl₃) δ –62.89; IR (film) 3054, 2987, 1724, 1517, 1446, 1325, 1139, 1112, 1012, 854 cm⁻¹; HRMS (EI) m/z calcd for C₂₀H₁₄F₃N₃O₄, 417.0936; found, 417.0936.

Ethyl 5-(Naphthalen-2-yl)-3-(4-nitrophenyl)pyrazine-2-carboxy*late (3ak)*. Yield 39.9 mg (50%), $R_f = 0.2$ (EtOAc:Hx = 1:5); white solid, melting point, $165-170$ °C; ¹H NMR (400 MHz, CDCl₃) δ 9.27 (s, 1H), 8.65 (d, $J = 1.1$ Hz, 1H), 8.39 (d, $J = 8.8$ Hz, 2H), 8.25 (dd, J = 8.6, 1.8 Hz, 1H), 8.03−8.00 (m, 2H), 7.93−7.89 (m, 3H), 7.62−7.56 (m, 2H), 4.38 (q, ^J = 7.1 Hz, 2H), 1.28 (t, ^J = 7.1 Hz, 3H); 13C{1 H} NMR (100 MHz, CDCl3) δ 165.6, 153.0, 151.5, 148.4, 144.1, 141.6, 140.0, 134.5, 133.3, 132.1, 130.0, 129.2, 129.0, 127.9, 127.88, 127.0, 123.9, 123.6, 62.5, 14.0; IR (film) 3057, 2982, 1730, 1520, 1434, 1348, 1138, 1013, 853 cm⁻¹; HRMS (EI) *m/z* calcd for C₂₃H₁₇N₃O₄, 399.1219; found, 399.1218.

Ethyl 3-(4-Nitrophenyl)-5-(thiophen-3-yl)pyrazine-2-carboxylate (3al). Yield 38.4 mg (54%), $R_f = 0.23$ (EtOAc:Hx = 1:5); pale yellow solid, melting point, 171–175 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.99 (s, 1H), 8.36 (d, J = 8.8 Hz, 2H), 8.18 (dd, J = 3.0, 1.2 Hz, 1H), 7.84 (d, J = 8.8 Hz, 2H), 7.79 (dd, J = 5.1, 1.2 Hz, 1H), 7.50 (dd, J = 5.1, 3.0 Hz, 1H), 4.35 (q, J = 7.1 Hz, 2H), 1.26 (t, J = 7.1 Hz, 3H);
¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.4, 151.7, 149.2, 148.3, 144.0, 141.1, 139.6, 137.8, 129.9, 127.6, 127.3, 126.0, 123.5, 62.5, 13.9; IR (film) 3073, 2984, 1710, 1521, 1460, 1351, 1140, 1015, 853 cm[−]¹ ; HRMS (EI) m/z calcd for C₁₇H₁₃N₃O₄S, 355.0627; found, 355.0626.

Ethyl 5-(Cyclohex-1-en-1-yl)-3-(4-nitrophenyl)pyrazine-2-carboxylate (3am). Yield 36 mg (51%), $R_f = 0.31$ (EtOAc:Hx = 1:5); yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 8.77 (s, 1H), 8.33 (d, J = 8.9 Hz, 2H), 7.79 (d, J = 8.9 Hz, 2H), 7.03−7.00 (m, 1H), 4.33 (q, J = 7.1 Hz, 2H), 2.60−2.57 (m, 2H), 2.37−2.33 (m, 2H), 1.86−1.80 (m, 2H), 1.75−1.69 (m, 2H), 1.25 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.7, 154.4, 150.7, 148.2, 144.4, 140.6, 138.7, 134.5, 134.0, 129.9, 123.4, 62.3, 26.3, 25.2, 22.4, 21.7, 13.9; IR (film) 3051, 2931, 1730, 1520, 1444, 1348, 1135, 1013, 852 cm⁻¹; HRMS (EI) *m/z* calcd for $C_{19}H_{19}N_3O_4$, 353.1376; found, 353.1373.

Ethyl 5-Butyl-3-(4-nitrophenyl)pyrazine-2-carboxylate (3an). Yield 32.9 mg (50%), $R_f = 0.36$ (EtOAc:Hx = 1:5); yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 8.56 (s, 1H), 8.34 (d, J = 8.8 Hz, 2H), 7.77 (d, J = 8.8 Hz, 2H), 4.33 (q, J = 7.1 Hz, 2H), 2.97−2.93 (m, 2H), 1.83−1.75 (m, 2H), 1.49−1.39 (m, 2H), 1.25 (t, J = 7.1 Hz, 3H), 0.97 $(t, J = 7.3 \text{ Hz}, 3\text{H})$; ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.7, 159.5, 151.2, 148.3, 144.1, 142.6, 141.4, 129.8, 123.5, 62.4, 35.3, 31.3, 22.4, 13.9, 13.8; IR (film) 3046, 2958, 1733, 1521, 1445, 1348, 1129, 1014, 853 cm⁻¹; HRMS (EI) *m/z* calcd for C₁₇H₁₉N₃O₄, 329.1376; found, 329.1377.

Ethyl 5-(3-Chlorophenyl)-3-phenylpyrazine-2-carboxylate (3bf). Yield 35.2 mg (52%), $R_f = 0.14$ (EtOAc:Hx = 1:10); yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 9.00 (s, 1H), 8.18–8.17 (m, 1H), 8.03– 8.00 (m, 1H), 7.74–7.70 (m, 2H), 7.53–7.45 (m, 5H), 4.32 (q, J = 7.1 Hz, 2H), 1.20 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.4, 153.0, 151.1, 142.9, 138.5, 137.3, 137.2, 135.4, 130.6, 130.4, 129.7, 128.8, 128.5, 127.5, 125.4, 62.2, 13.8; IR (film) 3064, 2981, 1732, 1421, 1139, 1064, 1022, 838 cm⁻¹; HRMS (EI) m/z calcd for $C_{19}H_{15}CIN_2O_2$, 338.0822; found, 338.0824.

Ethyl 5-(3-chlorophenyl)-3-(4-chlorophenyl)pyrazine-2-carboxy*late (3cf)*. Yield 54.5 mg (73%), $R_f = 0.17$ (EtOAc:Hx = 1:5); yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 9.01 (s, 1H), 8.15–8.14 (m, 1H), 8.01−7.99 (m, 1H), 7.66 (d, J = 8.5 Hz, 2H), 7.52−7.46 (m, 4H), 4.35 $(q, J = 7.1 \text{ Hz}, 2\text{H})$, 1.26 $(t, J = 7.1 \text{ Hz}, 3\text{H})$; ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.1, 151.9, 151.2, 142.6, 138.7, 137.0, 136.1, 135.7, 135.4, 130.8, 130.4, 130.2, 128.8, 127.5, 125.4, 62.4, 13.9; IR (film) 3068, 2981, 1733, 1444, 1139, 1067, 1013, 844 cm⁻¹; HRMS (EI) *m/z* calcd for $C_{19}H_{14}Cl_2N_2O_2$, 372.0432; found, 372.0431.

Ethyl 3-(3-Bromophenyl)-5-(3-chlorophenyl)pyrazine-2-carboxy*late (3df)*. Yield 59.1 mg (71%), $R_f = 0.37$ (EtOAc:Hx = 1:5); yellow

solid, melting point, 55−59 °C; ¹H NMR (400 MHz, CDCl₃) δ 9.02 $(s, 1H)$, 8.15–8.14 (m, 1H), 8.01 (dt, J = 6.7, 1.9 Hz, 1H), 7.86 (t, J = 1.8 Hz, 1H), 7.65−7.61 (m, 2H), 7.52−7.46 (m, 2H), 7.37 (t, J = 7.9 Hz, 1H), 4.35 (q, J = 7.1 Hz, 2H), 1.25 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.9, 151.5, 151.3, 142.8, 139.2, 139.0, 137.0, 135.4, 132.7, 131.8, 130.8, 130.5, 130.0, 127.5, 127.4, 125.5, 122.5, 62.4, 13.9; IR (film) 3066, 2980, 1732, 1422, 1141, 1068, 1139, 1012, 841 cm⁻¹; HRMS (EI) m/z calcd for C₁₉H₁₄BrClN₂O₂, 415.9927; found, 415.9926.

Ethyl 3-(4-Bromophenyl)-5-(3-chlorophenyl)pyrazine-2-carboxy*late (3ef)*. Yield 60.1 mg (72%), $R_f = 0.18$ (EtOAc:Hx = 1:5); white solid, melting point, 82−87 °C; ¹H^{NMR} (400 MHz, CDCl₃) δ 9.01 (s, 1H), 8.15−8.14 (m, 1H), 8.01−7.99 (m, 1H), 7.65 (d, J = 8.6 Hz, 2H), 7.59 (d, J = 8.6 Hz, 2H), 7.52−7.46 (m, 2H), 4.35 (q, J = 7.1 Hz, 2H), 1.26 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.1, 152.0, 151.2, 142.5, 138.8, 137.0, 136.2, 135.4, 131.7, 130.8, 130.44, 130.41, 127.5, 125.4, 124.5, 62.4, 13.9; IR (film) 3067, 2981, 1732, 1443, 1140, 1069, 1034, 1009, 842 cm⁻¹; HRMS (EI) m/z calcd for $C_{19}H_{14}BrClN_2O_2$, 415.9927; found, 415.9928.

Ethyl 5-(3-Chlorophenyl)-3-(4-(methoxycarbonyl)phenyl) pyrazine-2-carboxylate (3ff). Yield 55.6 mg (70%), $R_f = 0.29$ (EtOAc:Hx = 1:5); white solid, melting point, 142-147 °C; ¹H NMR (400 MHz, CDCl₃) δ 9.04 (s, 1H), 8.18 (d, J = 8.6 Hz, 2H), 8.16−8.15 (m, 1H), 8.03−8.00 (m, 1H), 7.78 (d, J = 8.6 Hz, 2H), 7.52−7.47 (m, 2H), 4.33 (q, J = 7.2 Hz, 2H), 3.97 (s, 3H), 1.21 (t, J = 7.2 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.6, 165.9, 152.2, 151.3, 142.8, 141.6, 139.1, 137.0, 135.5, 131.1, 130.9, 130.5, 129.7, 128.9, 127.6, 125.4, 62.4, 52.4, 13.8; IR (film) 3067, 2983, 1725, 1435, 1279, 1140, 1067, 1016, 862 cm⁻¹; HRMS (EI) m/z calcd for $C_{21}H_{17}CIN_2O_4$, 396.0877; found, 396.0878.

Ethyl 5-(3-Chlorophenyl)-3-(naphthalen-2-yl)pyrazine-2-carboxylate (3gf). Yield 45.1 mg (58%), $R_f = 0.26$ (EtOAc:Hx = 1:5); yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 9.02 (s, 1H), 8.22–8.20 (m, 2H), 8.06−8.03 (m, 1H), 7.99−7.91 (m, 3H), 7.83 (dd, J = 8.5, 1.8 Hz, 1H), 7.59−7.53 (m, 2H), 7.52−7.47 (m, 2H), 4.32 (q, J = 7.1 Hz, 2H), 1.15 (t, J = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.5, 152.9, 151.2, 143.1, 138.5, 137.3, 135.4, 134.6, 133.8, 133.0, 130.7, 130.4, 128.8, 128.7, 128.3, 127.8, 127.6, 127.2, 126.7, 125.9, 125.5, 62.3, 13.9; IR (film) 3061, 2980, 1731, 1445, 1139, 1068, 1014, 861 cm⁻¹; HRMS (EI) *m/z* calcd for C₂₃H₁₇ClN₂O₂, 388.0979; found, 388.0977.

Ethyl 3-(4-Nitrophenyl)-5-phenyl-1-tosyl-1,4-dihydropyrazine-2 carboxylate (4). $R_f = 0.35$ (EtOAc:Hx = 1:2); yellow solid, melting point, 224−227 °C; ¹H NMR (400 MHz, acetone- d_6) δ 11.13 (s, 1H), 8.26 (d, J = 9.0 Hz, 2H), 7.86 (d, J = 9.0 Hz, 2H), 7.83–7.79 (m, 3H), 7.42 (d, J = 8.3 Hz, 2H), 7.36−7.32 (m, 2H), 7.29−7.25 (m, 1H), 7.17 $(d, J = 8.0 \text{ Hz}, 2\text{H})$, 3.92 $(q, J = 7.1 \text{ Hz}, 2\text{H})$, 2.33 $(s, 3\text{H})$, 1.11 $(t, J =$ 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, acetone- d_6) δ 164.6, 148.1, 144.0, 139.0, 137.8, 133.1, 131.9, 131.7, 131.0, 129.8, 129.0, 128.4, 128.2, 128.1, 123.7, 119.5, 113.0, 60.7, 21.4, 14.2; IR (film) 3280, 3064, 2982, 1685, 1516, 1400, 1343, 1256, 1161 cm[−]¹ ; HRMS (EI) m/ z calcd for $C_{26}H_{23}N_3O_6S$, 505.1308; found, 505.1310.

Ethyl 3-(4-Nitrophenyl)-5-phenyl-1-tosyl-1,2-dihydropyrazine-2 carboxylate (5). $R_f = 0.34$ (EtOAc:Hx = 1:5); yellow oil; ¹H NMR $(400 \text{ MHz}, 400 \text{ MHz}, \text{CDCl}_3) \delta 8.29 \text{ (d, } J = 9.0 \text{ Hz}, 2H)$, 8.10 (d, $J =$ 9.0 Hz, 2H), 7.76−7.74 (m, 2H), 7.60 (d, J = 8.4 Hz, 2H), 7.43−7.39 $(m, 2H)$, 7.37–7.32 $(m, 1H)$, 7.14 $(d, J = 8.0 \text{ Hz}, 2H)$, 7.10 $(d, J = 1.4 \text{ Hz})$ Hz, 1H), 6.05 (d, J = 1.4 Hz, 1H), 4.18−3.99 (m, 2H), 2.33 (s, 3H), 1.11 (t, J = 7.1 Hz, 3H) ; ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.0, 148.8, 145.1, 145.0, 141.2, 135.5, 135.2, 135.0, 129.9, 128.7, 128.5, 128.3, 126.6, 124.9, 123.7, 111.4, 62.8, 52.8, 21.6, 13.9; IR (film) 3081, 2981, 1733, 1596, 1521, 1346, 1317, 1169 cm[−]¹ ; HRMS (EI) m/z calcd for $C_{26}H_{23}N_3O_6S$, 505.1308; found, 505.1308.

■ ASSOCIATED CONTENT

S Supporting Information

Characterization data, X-ray crystallography data (3ab, cif), and H, $^{13}C(^{1}H)$, and ^{19}F NMR spectra for new compounds. This

material is available free of charge via the Internet at http:// pubs.acs.org.

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Notes

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■ REFERENCES

(1) Sato, N. In Comprehensive Heterocyclic Chemistry II; Katritzky, A. R., Rees, C. W., Boulton, A. J., Eds.; Elsevier: Oxford, 1996.

(2) (a) Urban, S.; Hickford, S. J. H.; Blunt, J. W.; Munro, M. H. G. Curr. Org. Chem. 2000, 4, 765. (b) Ohta, A.; Aoyagi, Y. Rev. Heteroat. Chem. 1998, 18, 141. (c) Gohlke, H.; Gundisch, D.; Schwarz, S.; Seitz, G.; Tilotta, M. C.; Wegge, T. J. Med. Chem. 2002, 45, 1065. (d) Cavalier, J. F.; Burton, M.; Dussart, F.; Marchand, F.; Rees, J. F.; Marchant-Brynaert, J. Bioorg. Med. Chem. 2001, 9, 1037. (e) Fruit, C.; Turck, A.; Plé, N.; Mojovic, L.; Quéguiner, G. Tetrahedron 2001, 57, 9429. (f) Hirsh, A. J.; Molino, B. F.; Zhang, J.; Astakhova, N.; Geiss, W. B.; Sargent, B. J.; Swenson, B. D.; Usyatinsky, A.; Wyle, M. J.; Boucher, R. C.; Smith, R. T.; Zamurs, A.; Johnson, M. R. J. Med. Chem. 2006, 49, 4098. (g) Berkers, C. R.; Leestemaker, Y.; Schuurman, K. G.; Ruggeri, B.; Jones-Bolin, S.; Williams, M.; Ovaa, H. Mol. Pharmaceutics 2012, 9, 1126. (h) Wu, X.-A.; Zhao, Y.-M.; Yu, N.-J. J. Asian Nat. Prod. Res. 2007, 9, 437. (i) Duran, R.; Zubia, E.; Ortega, M. J.; Naranjo, S.; Salva, J. Tetrahedron 1999, 55, 13225.

(3) Bü chi, G.; Galindo, J. J. Org. Chem. 1991, 56, 2605.

(4) Aparicio, D.; Attanasi, O. A.; Filippone, P.; Ignacio, R.; Lillini, S.; Mantellini, F.; Palacios, F.; de los Santos, J. M. J. Org. Chem. 2006, 71, 5897.

(5) Elmssty, T. A.; Castle, L. Org. Lett. 2005, 7, 5529.

(6) Matsushita, H.; Lee, S.-H.; Yoshida, K.; Clapham, B.; Koch, G.; Zimmermann, J.; Janda, K. D. Org. Lett. 2004, 6, 4627.

(7) Taber, D. F.; DeMatteo, P. W.; Taluskie, K. V. J. Org. Chem. 2007, 72, 1492.

(8) Gnanaprakasam, B.; Balaraman, E.; Ben-David, Y.; Milstein, D. Angew. Chem., Int. Ed. 2011, 50, 12240.

(9) Palacios, F.; de Retana, A. M. O.; Gil, J. I.; de Munain, R. L. Org. Lett. 2002, 4, 2405.

(10) (a) Mehta, V. P.; Modha, S. G.; Van der Eycken, E. V. J. Org. Chem. 2010, 75, 976. (b) Modha, S. G.; Trivedi, J. C.; Mehta, V. P.; Ermolat'ev, D. S.; Van der Eycken, E. V. J. Org. Chem. 2011, 76, 846. (c) Mehta, V. P.; Sharma, A.; Van Hecke, K.; Van Meervelt, L.; Van der Eycken, E. V. J. Org. Chem. 2008, 73, 2382. (d) Mosrin, M.; Bresser, T.; Knochel, P. Org. Lett. 2009, 11, 3406. (e) Liu, W.; Wise, D. S.; Townsend, L. B. J. Org. Chem. 2001, 66, 4783. (f) Buron, F.; Plé, N.; Turck, A.; Queguiner, G. J. Org. Chem. 2005, 70, 2616.

(11) For reviews, see: (a) Davies, H. M. L.; Alford, J. S. Chem. Soc. Rev. 2014, 43, 5151. (b) Anbarasan, P.; Yadagiri, D.; Rajasekar, S. Synthesis 2014, 46, 3004 and references therein. (c) Horneff, T.; Chuprakov, S.; Chernyak, N.; Gevorgyan, V.; Fokin, V. V. J. Am. Chem. Soc. 2008, 130, 14972. (d) Miura, T.; Yamauchi, M.; Murakami, M. Chem. Commun. 2009, 1470. (e) Chattopadhyay, B.; Gevorgyan, V. Org. Lett. 2011, 13, 3746. (f) Shi, Y.; Gevorgyan, V. Org. Lett. 2013, 15, 5394. (g) Miura, T.; Hiraga, K.; Biyajima, T.; Nakamuro, T.; Murakami, M. Org. Lett. 2013, 15, 3298. (h) Schultz, E. E.; Sarpong, R. J. Am. Chem. Soc. 2013, 135, 4696. (i) Chuprakov, S.; Kwok, S. W.; Fokin, V. V. J. Am. Chem. Soc. 2013, 135, 4652. (j) Parr, B. T.; Green, S. A.; Davies, H. M. L. J. Am. Chem. Soc. 2013, 135, 4716. (k) Zibinsky, M.; Fokin, V. V. Angew. Chem., Int. Ed. 2013, 52, 1507. (l) Spangler, J. E.; Davies, H. M. L. J. Am. Chem. Soc. 2013, 135, 6802. (m) Miura, T.;

Funakoshi, Y.; Murakami, M. J. Am. Chem. Soc. 2014, 136, 2272. (n) Kwok, S. W.; Zhang, L.; Grimster, N. P.; Fokin, V. V. Angew. Chem., Int. Ed. 2014, 53, 3452. (o) Zibinsky, M.; Fokin, V. V. Org. Lett. 2011, 13, 4870. (p) Alford, J. S.; Spangler, J. E.; Davies, H. M. L. J. Am. Chem. Soc. 2013, 135, 11712. (q) Jeon, H. J.; Jung, D. J.; Kim, J. H.; Kim, Y.; Bouffard, J.; Lee, S.-g. J. Org. Chem. 2014, 79, 9865. (r) Jung, D. J.; Jeon, H. J.; Kim, J. H.; Kim, Y.; Lee, S.-g. Org. Lett. 2014, 16, 2208. (s) Chattopadhyay, B.; Gevorgyan, V. Angew. Chem., Int. Ed. 2012, 51, 862. (t) Gulevich, A. V.; Gevorgyan, V. Angew. Chem., Int. Ed. 2013, 52, 1371. (u) Tang, X.-Y.; Zhang, Y.-S.; He, L.; Wei, Y.; Shi, M. Chem. Commun. 2015, 51, 133. (v) Liu, R.; Zhang, M.; Winston-McPherson, G.; Tang, W. Chem. Commun. 2013, 49, 4376.

(12) (a) Kim, S.; Mo, J.; Kim, J.; Ryu, T.; Lee, P. H. Asian J. Org. Chem. 2014, 3, 926. (b) Kim, C.-E.; Park, S.; Eom, D.; Seo, B.; Lee, P. H. Org. Lett. 2014, 16, 1900. (c) Kim, C.-E.; Park, Y.; Park, S.; Lee, P. H. Adv. Synth. Catal. 2014, 357, 210. (d) Seo, B.; Jeon, W.; Kim, J.; Kim, S.; Lee, P. H. J. Org. Chem. 2015, 80, 722.

(13) Park, S.; Yong, W.-S.; Kim, S.; Lee, P. H. Org. Lett. 2014, 16, 4468.

(14) During the preparation of this manuscript, Zhang et al. reported synthesis of dihydropyrazines through the Rh-catalyzed transannulation of 1-sulfonyl-1,2,3-triazoles with 2H-azirines: Ding, H.; Hong, S.; Zhang, N. Tetrahedron Lett. 2015, 56, 507. After submission of this paper, Park et al. reported synthesis of unsymmetrical pyrazines based on α -diazo oximine ethers: Loy, N. S.; Kim, S.; Park, C.-M. Org. Lett. 2015, DOI: 10.1021/ol5034173.

(15) (a) Yoo, E. J.; Ahlquist, M.; Kim, S. H.; Bae, I.; Fokin, V. V.; Sharpless, K. B.; Chang, S. Angew. Chem., Int. Ed. 2007, 46, 1730. (b) Raushel, J.; Fokin, V. V. Org. Lett. 2010, 12, 4952. (c) Liu, Y.; Wang, X.; Xu, J.; Zhang, Q.; Zhao, Y.; Hu, Y. Tetrahedron 2011, 67, 6294.

(16) (a) Wei, Y. J.; Chan, W. C.; Park, C.-M. J. Am. Chem. Soc. 2012, 134, 4104. (b) Qi, X.; Xu, X.; Park, C.-M. Chem. Commun. 2012, 48, 3996. (c) Loy, N. S. Y.; Singh, A.; Xu, X.; Park, C.-M. Angew. Chem., Int. Ed. 2013, 53, 2212.

(17) CCDC 1037286 (3ab) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Center via www.ccdc.cam. ac.uk/data_request/cif.

(18) In reference 14, authors suggested an intramole[cular hydrogen](www.ccdc.cam.ac.uk/data_request/cif) [transfer mechanism.](www.ccdc.cam.ac.uk/data_request/cif)